# 7-Oxaspiro[5.9]pentadecane-1,8,13-trione 

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#### Abstract

The crystal structure of the title compound, $\mathrm{C}_{14} \mathrm{H}_{20} \mathrm{O}_{4}$, a macrocyclic lactone, is composed of independent molecules with normal molecular dimensions. The average bond distances are $\mathrm{C}_{s p^{2}}-\mathrm{C}_{s p_{3}^{3}} 1.526(6), \mathrm{C}_{s p^{3}}-\mathrm{C}_{s p^{2}}$ 1.514 (14) and $\mathrm{C}=\mathrm{O} 1.210$ (3) $\AA$. The six-membered ring adopts a chair conformation, while the tenmembered heterocyclic ring is in a boat-chair-boat (bcb) conformation. There are no unusual interactions between the molecules which are separated by normal van der Waals distances.


## Comment

Cyclohexanone, together with its derivatives, are useful building blocks for the construction of organic compounds, especially natural products. There are several macrocyclic lactones derived from fermentation which are used as antibiotics and so have clinical importance, e.g. erythromycine, rapamycine and FK-506, etc. (Caufield, 1995). Rapamycine and FK-506 became the drugs of choice in the late 1980s as the two new classes of immunosuppressants having the promise of supplementing cyclosporin A for the treatment of transplant rejection, and hopefully, also having use in the treatment of autoimmune disease (Caufield, 1995).

Continuing our studies on the preparation of biologically active compounds derived from Mannich bases (Parvez et al., 1998), we have synthesized the title compound, (1), a macrocyclic lactone.

(1)

The crystal structure of (1) is composed of a tenmembered heterocyclic ring ( $\mathrm{O} 1, \mathrm{Cl}-\mathrm{C} 9$ ), with two ketonic groups (at Cl and C 6 ), and a six-membered hydrocarbon ring (C9-C14), with a ketonic function at C 10 ; the two rings are linked together at C 9 .

The ten-membered ring has a boat-chair-boat (bcb) conformation, while the six-membered ring adopts a classical chair conformation, with torsion angles in the range $\pm\left[53.9(2)-60.0(2)^{\circ}\right]$. The structure contains discrete molecules of (1) separated by normal van der Waals distances. The molecular dimensions are normal and lie within expected values for corresponding bond distances and angles (Orpen et al., 1994), with mean bond distances $\mathrm{C}_{s p^{3}}-\mathrm{C}_{s p^{3}}{ }^{1.526(6)} \mathrm{C}_{s p^{3}}-\mathrm{C}_{s p^{2}}$ 1.514 (14) and $\mathrm{C}=\mathrm{O} 1.210$ (3) A , while the $\mathrm{C}_{s p^{3}}-\mathrm{O}$ and $\mathrm{C}_{s p^{2}}-\mathrm{O}$ distances are 1.471 (2) and $1.349(2) \AA$, respectively.

Some of the compounds containing a similar heterocyclic ring which have been studied by crystallographic methods are oxacyclodeca-2,6-dione (Fedeli \& Dunitz, 1968), thiobis(cephalosporolide A) (Mabelis et al., 1981), cephalosporolide C and cephalosporolide B 3-O-methanesulfonate (Ackland et al., 1985), endo,endo,exo-2,6,10-tribromo-exo-5-methoxy-13-oxa-trans-bicyclo[7.3.1]tridecane (Rissanen \& Haufe, 1988), ( $9 R, 11 R$ )-9,11-epoxy-14,15,17-trinor-8,9-secolabdan-8,13olide (Grant et al., 1991) and nonanolactone (Wiberg et al., 1991).


Fig. 1. ORTEPII (Johnson, 1976) drawing of (1) with the atomic numbering scheme. Displacement ellipsoids are plotted at the $50 \%$ probability level and H atoms have been assigned arbitrary radii.

## Experimental

3-Hydroxy-2,16-dioxatetracyclohexadecane (Parvez et al., 1998) ( $1.2 \mathrm{~g}, 0.005 \mathrm{~mol}$ ) in acetic acid ( 2.0 ml ) was treated with $\mathrm{CrO}_{3}(1.0 \mathrm{~g}, 0.01 \mathrm{~mol})$ dissolved in acetic acid ( 3.0 ml ) and water ( 1.5 ml ). The mixture was shaken vigorously and then allowed to stand for 18 h at room temperature, whereupon it was poured into an ice-sodium bicarbonate mixture and extracted three times with ether. The combined ethereal layers were washed with water and dried over anhydrous sodium sulfate. Removal of the solvent in vacuo and recrystallization from ethyl acetate afforded (1) as large colorless needles, m.p. $402-403 \mathrm{~K}(0.44 \mathrm{~g}, 40 \%)$.

## Crystal data

$\mathrm{C}_{14} \mathrm{H}_{20} \mathrm{O}_{4}$
Mo $K \alpha$ radiation
$\lambda=0.71069 \AA$

Monoclinic
$P 2_{1} / n$
$a=11.667$ (3) $\AA$
$b=7.292(2) \AA$
$c=15.275(3) \AA$
$\beta=97.02(2)^{\circ}$
$V=1289.8(5) \AA^{3}$
$Z=4$
$D_{x}=1.299 \mathrm{Mg} \mathrm{m}^{-3}$
$D_{m}$ not measured

## Data collection

| Rigaku AFC- $6 S$ diffractom- | $R_{\text {int }}=0.046$ |
| :--- | :--- |
| $\quad$ eter | $\theta_{\text {max }}=25.0^{\circ}$ |
| $\omega / 2 \theta$ scans | $h=0 \rightarrow 13$ |
| Absorption correction: none | $k=0 \rightarrow 8$ |
| 2395 measured reflections | $l=-18 \rightarrow 18$ |
| 2279 independent reflections | 3 standard reflections |
| 1662 reflections with | every 200 reflections |
| $I>2 \sigma(I)$ | intensity decay: none |

## Refinement

Refinement on $F^{2}$
$R(F)=0.037$
$w R\left(F^{2}\right)=0.107$
$S=1.171$
2273 reflections
165 parameters
H atoms riding
$w=1 /\left[\sigma^{2}\left(F_{o}^{2}\right)+(0.05 P)^{2}\right.$
$+0.65 P]$
where $P=\left(F_{o}^{2}+2 F_{c}^{2}\right) / 3$
H atoms were included at geometrically idealized positions, with a C-H distance of $0.95 \AA$.

Data collection: MSC/AFC Diffractometer Control Software (Molecular Structure Corporation, 1988). Cell refinement: MSC/AFC Diffractometer Control Software. Data reduction: TEXSAN (Molecular Structure Corporation, 1994). Program(s) used to solve structure: SAPI91 (Fan, 1991). Program(s) used to refine structure: SHELXL93 (Sheldrick, 1993). Molecular graphics: TEXSAN. Software used to prepare material for publication: SHELXL93.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: FG1408). Services for accessing these data are described at the back of the journal.

## References

Ackland, M. J., Hanson, J. R., Hitchcock, P. B. \& Ratcliffe, A. H. (1985). J. Chem. Soc. Perkin Trans. pp. 843-847.

Caufield, C. E. (1995). Curr. Pharmaceut. Des. pp. 145-160.
Fan, H.-F. (1991). SAPI91. Structure Analysis Program with Intelligent Control. Rigaku Corporation. Tokyo, Japan.
Fedeli, W. \& Dunitz, J. D. (1968). Helv. Chim. Acta, 51. 445-458.
Grant, P. K., Hanton, L. R., Lynch, G. P., Simpson, J. \& Slim, G. C. (1991). Aust. J. Chem. 44, 897-906.
$(\Delta / \sigma)_{\text {max }}<0.001$
$\Delta \rho_{\text {max }}=0.222 \mathrm{e}^{\AA^{-3}}$
$\Delta \rho_{\text {min }}=-0.235$ e $\AA^{-3}$
Extinction correction: none
Scattering factors from International Tables for Crystallography (Vol. C)

Cell parameters from 25 reflections
$\theta=18-25^{\circ}$
$\mu=0.094 \mathrm{~mm}^{-1}$
$T=200(1) \mathrm{K}$
Block, cut from a large needle
$0.45 \times 0.38 \times 0.30 \mathrm{~mm}$
Colorless

$$
\begin{aligned}
& R_{\text {int }}=0.046 \\
& \theta_{\text {max }}=25.0^{\circ} \\
& h=0 \rightarrow 13 \\
& k=0 \rightarrow 8 \\
& l=-18 \rightarrow 18 \\
& 3 \text { standard reflections } \\
& \quad \text { every } 200 \text { reflections } \\
& \text { intensity decay: none }
\end{aligned}
$$

Johnson, C. K. (1976). ORTEPII. Report ORNL-5I38. Oak Ridge National Laboratory, Tennessec, USA.
Mabelis, R. P., Ratcliffe, A. H., Ackland. M. J.. Hanson, J. R. \& Hitchcock, P. B. (1981). J. Chem. Soc. Chem. Commun. pp. 10061007.

Molecular Structure Corporation (1988). MSC/AFC Diffractometer Control Software. MSC, 3200 Research Forest Drive. The Woodlands, TX 77381. USA.
Molecular Structure Corporation (1994). TEXSAN. Single Crystal Structure Analysis Sofinare. MSC. 3200 Research Forest Drive, The Woodlands. TX 77381, USA.
Orpen, A. G.. Brammer. L., Allen, F. H., Kennard. O.. Watson, D. G. \& Taylor, R. (1994). Structure Correlation. Vol. 2, edited by H.-B. Bürgi \& J. D. Dunitz, pp. 751-858. New York: VCH.
Parvez. M., Sultana, N., Sarfaraz, T. B. \& Husain. S. A. (1998). Acta Cryst. C54. In the press.
Rissanen, K. \& Haufe, G. (1988). Acta Cnst. C44, 1803-1805.
Sheldrick, G. M. (1993). SHELXL93. Program for the Refinement of Crustal Structures. University of Götuingen. Germany.
Wiberg, K. W., Waldron, R. F., Schulte, G. \& Saunders. M. (1991). J. Am. Chem. Soc. 113, 971-977.

Acta Cryst. (1998). C54, 790-792

## 14-O-Benzoyl-8-ethoxybikhaconine and 14-O-Benzoyl-8-methoxybikhaconine

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## Abstract

The molecular structures of two new $\mathrm{C}_{19}$ norditerpenoid alkaloids, 14-O-benzoyl-8-methoxybikhaconine $[(1 \alpha, 6 \alpha, 14 \alpha, 16 \beta)$-20-ethyl-13-hydroxy-1,6,8,16-tetramethoxy-4-methoxymethylaconitan-14-yl benzoate, $\mathrm{C}_{33} \mathrm{H}_{47} \mathrm{NO}_{8}$, (1)] and 14-O-benzoyl-8-ethoxybikhaconine [( $1 \alpha, 6 \alpha, 14 \alpha, 16 \beta)$-8-ethoxy-20-ethyl-13-hydroxy-1,6,16-trimethoxy-4-methoxymethylaconitan-14-yl benzoate, $\mathrm{C}_{34} \mathrm{H}_{49} \mathrm{NO}_{8}$, (2)], isolated for the first time from the roots of Aconitum chasmanthum Stapf ex Holmes of Pakistani origin, have been determined. The two alkaloids differ by one $\mathrm{CH}_{2}$ moiety in a side chain (8-methoxy versus 8-ethoxy), co-crystallize in a 65 (2):35 (2) ratio, i.e. $0.65 \mathrm{C}_{33} \mathrm{H}_{47} \mathrm{NO}_{8} \cdot 0.35 \mathrm{C}_{34} \mathrm{H}_{49} \mathrm{NO}_{8}$, and are inseparable by thin-layer and column chromatography. The conformations of the rings in the two alkaloids are: $A$ and $E$, chairs; $D$, half-chair; $C$ and $F$, envelopes; and $B$, boat. The molecular dimensions are normal; the mean bond distances are $\mathrm{C}_{s p}:-\mathrm{N} 1.463(6)$, $\mathrm{C}_{s p^{3}}-\mathrm{C}_{s p^{3}} 1.54(2), \mathrm{C}_{s p^{3}}-\mathrm{O} 1.41(2), \mathrm{C}_{s p^{2}}-\mathrm{O} 1.342(3)$ and $\mathrm{C}=\mathrm{O} 1.204(4) \mathrm{A}$. There is a short intramolecular

